Marine Natural Products: 9α ,11 α -Epoxycholest-7-ene- 3β , 5α , 6β ,19-tetrol 6-Acetate from a Sponge, Dysidea sp.

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Numerous new sterols have been isolated from marine organisms, most having novel side-chain alkylation patterns but also a few with nuclear modifications.¹ A second, as yet small class of unusual sterols are the polyhydroxy and ring-C seco sterols;1b however, one of us has noted earlier that future studies would be likely to uncover additional polar, hydroxylated sterols in marine organisms.^{1b} In this paper we report the isolation of a new hydroxylated sterol with the unusual features of a 9,11-epoxide and a It has been noted² that 19-19-hydroxyl group. hydroxylated sterols may be intermediates in the formation of 19-demethylsterols isolated from marine organisms. 9,11-Epoxy sterols are logical intermediates in the formation of 9,11-seco sterols.

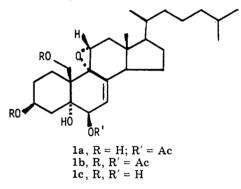
The new sterol was isolated from a Dysidea sp. of sponge collected in Guam. Silica gel followed by highpressure C_{18} reversed-phase chromatography of the chloroform:methanol (1:1) and methanol extracts yielded a crystalline solid: mp 229–230 °C, $[\alpha]^{26}_{D}$ +42.6° (c 0.07, CHCl₃). The formula $C_{29}H_{46}O_6$ was established by mass spectrometry: low-resolution M⁺ 490; high-resolution 472.31624 (C₂₉H₄₄O₅, calcd 472.31887, M⁺ - H₂O). Infrared data indicated the presence of hydroxyl groups (3585, 3450 cm^{-1}) and an acetate (1736, 1240 cm^{-1}). The ¹H NMR spectrum confirmed the presence of one acetate (δ 2.15 (s)) and revealed the presence of three hydroxyl groups (exchangeable signals at δ 5.82 (s), 6.21 (d), and 6.55 (t)). The ¹H NMR spectrum also contained signals for four of the five methyl groups typical of a sterol: δ 0.60 (s, C-18), 0.86 (6 H, d, J = 7 Hz, C-26, 27), and 0.92 (d, J = 7 Hz, C-21). Although the methyl singlet typical for the C-19 protons was absent, an AB quartet centered at δ 3.90, J = 12 Hz was observed, which was ascribable² to a hydroxylated C-19 group. Consistent with this assignment were prominent $(\sim 20\%)$ losses of CH₂OH + AcOH and CH₂OH + AcOH + H₂O in the mass spectrum.

In the mass spectrum, peaks were also observed at 299.16488 for M^+ - (CH₃CO₂H + C₈H₁₇) and 281.15433 M⁺ $-(H_2O + CH_3CO_2H + C_8H_{17})$ (see Experimental Section), indicating a conventional C_8H_{17} side chain.³ A peak was also observed at 245.11806, corresponding to loss of the side chain plus ring-D³ $[M^+ - (CH_3CO_2 + CH_2OH +$ $C_{11}H_{23}$), and thus it was established that all of the oxygen substituents were confined to the A, B, and/or C rings in the molecule.

The presence of a typical 3β -hydroxyl group was indicated by a broad methine multiplet at δ 4.06, the lower than normal chemical shift being consistent with deshielding from an axial 5α -hydroxyl group.⁴ This methine signal was shifted downfield to δ 4.75 in pyridine- d_5 , and irradiation of it collapsed a double doublet (1 H) at δ 2.77 (J = 14.2, 5.2 Hz) to a doublet (J = 14.2 Hz). This is as expected for a 4α -hydrogen next to a substituted C-5 postion in an A/B trans steroid, providing further evidence for C-5 substitution.

Two low-field signals with one common coupling constant, δ 5.22 (J = 3.5, 0.8 Hz) and 5.32 (J = 3.5, 1.0 Hz) were assigned to an acetoxy-deshielded allylic methine proton and a vicinal olefinic proton, respectively. Decoupling and difference decoupling confirmed that both of these protons were coupled (0.8 and 1.0 Hz, respectively) to a common allylic methine proton that resonated at δ 2.36 (br dd, J = 7.6, 6.6 Hz). These data require the fragment CHOAcCH=CCHCH₂.

The only remaining downfield signal was a doublet at δ 3.40 (J = 4.7 Hz), suitable for the proton of a trisubstituted epoxide, a ring that would account for the last of the seven degrees of unsaturation present in the new sterol. Decoupling established that the δ 3.40 proton was coupled to a proton resonating at δ 2.28 (J = 15.1, 4.7 Hz). The multiplicity of the δ 2.28 signal demonstrates that the corresponding proton is part of a methylene group next to a blocked position as in the partial structure >C(O)-CHCH₂. Incorporation of these last two partial structures into the sterol nucleus, keeping all oxygens in rings A, B, or C yields structure 1a. The rather high-field postion of the C-18 methyl signal in this structure is in good agreement with that expected⁵ for a Δ^7 sterol.



Acetylation of 1a yielded the triacetate 1b, confirming the presence of two acetylatable oxygens. Hydrolysis of 1a with KOH in aqueous ethanol afforded a product, 1c, whose low-resolution mass spectrum showed the expected M^+ of 448, appropriate for $C_{27}H_{44}O_5$. Product 1c lacked carbonyl absorption in the IR, thus removing any doubt that the original compound contained a second carbonyl absorption that was masked by the acetate carbonyl. In the ¹H NMR spectrum of 1c in pyridine- d_5 , only one proton appeared below δ 5 (δ 5.72), thus confirming that one of the $\delta > 5$ signals in 1a was due to an acetoxy-deshielded allylic methine proton.

Tori and co-workers⁶ have shown that for 9β , 11β -epoxy steroids $J_{11\alpha,12\alpha}$ and $J_{11\alpha,12\beta}$ are ~1.5 Hz while for the $9\alpha,11\alpha$ -isomer $J_{11\beta,12\alpha} \simeq 0, J_{11\beta,12\beta} \simeq 4.5$ Hz. Hence the single 4.7-Hz coupling for H-11 in 1a confirms the 9α , 11α stereochemistry. The 3.5-Hz coupling of H-6 to H-7 is compatible with the near-zero dihedral angle for these protons in the 6β -acetoxy- Δ^7 moiety, as assigned in 1a, whereas the epimeric 6α -acetoxy isomer would be expected to have a near-zero coupling for H-6/H-7.7

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therein.

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⁽⁵⁾ Kokke, W. C. M. C.; Bohlin, L.; Fenical W.; Djerassi, C. Phytochemistry 1982, 21, 881.

⁽⁶⁾ Tori, K.; Komeno, T.; Nakagawa, T. J. Org. Chem. 1964, 29, 1136. (7) Cf. $J_{6,7} < 1$ Hz in 24-methylenecholest-5-ene-3 β ,7 β ,19-triol; see ref

Only one other 19-hydroxylated sterol has been isolated² from natural sources to the best of our knowledge.

Sterol 1 was found to be slightly cytotoxic, ED_{50} in PS, 4.9 μ g/mL.⁸

Experimental Section

Melting points are uncorrected. Infrared spectra were taken on a Perkin-Elmer 298 spectrophotometer. NMR spectra were taken on Varian XL-100 and Nicolet 270 MHz instruments in the solvent specified; signals are reported in parts per million (δ) downfield from internal tetramethylsilane. Mass spectra were taken on CEC 110 (Du Pont) and Hewlett-Packard 5985B spectrometers. A Perkin-Elmer 141 polarimeter was used for obtaining optical rotations. The chromatographic adsorbent used was Brinkmann silica gel 60 (230-400 mesh). An Altex 5- μ m 10 mm \times 25 cm preparative silica gel (LiChrosorb 60) column was used for HPLC separations.

Isolation of 1a. Specimens of the sponge Dysidea sp. were collected at $\sim 10-15$ M around Guam Is. and transported frozen to Oklahoma. Freshly thawed material (wet weight 177.4 g) was extracted at room temperature first with chloroform-methanol (1:1) for 2 days and then with methanol for 1 week. Evaporation of the extracts in vacuo afforded residues weighing 1.64 and 0.36 g, respectively. Chromatography of 1.5 g of the combined extracts on 100 g of silica gel using chloroform followed by chloroform with increasing amounts of methanol afforded 18 fractions. Crystallization of fraction 13 (34 mg, eluted with chloroform-methanol (98:2)) from 4 mL of chloroform-hexane (1:2) overnight in a refrigerator yielded 6 mg of white crystals. This crystalline material was purified further by high-pressure liquid chromatography using a reverse-phase C_{18} column with a mobile phase of methanol-water (90:10). Crystallization of the major HPLC fraction from chloroform-hexane (1:2) yielded 4.2 mg of white crystals: mp 229–230 °C; $[\alpha]_{D}^{26}$ +42.6° (*c* 0.07, CHCl₃); IR(CHCl₃) 3585, 3650–3250 (brd), 1736, 1665 (vs), 1468, 1375, 1240 cm⁻¹; 270-MHz ¹H NMR (CDCl₃) (number of protons, multiplicity, J in Hz, assignment) δ 0.60 (3, s, H-18), 0.86 (6, d, J = 7 Hz, H-26, 27), 0.92 (3, d, J = 7 Hz, H-21), 2.07 (1, dd, J = 14.2, 5.2 Hz, 4α -H), 2.15 (3, s, OAc), 2.18 (1, dd, J = 15.0, 6.3 Hz, 12β -H), 2.36 (1, m, 14 α -H), 3.40 (1, d, J = 4.7 Hz, 11 β -H), 3.80 (1, dd, J = 12, 4.35Hz, H-19 coupled also with OH), 6-H), (1, dd, J = 12, 4.35 Hz,H-19 coupled also with OH), 4.06 (1, m, 3α -H), 5.22 (1, dd, J = 3.5, 1.0 Hz, H-7), 5.32 (1, dd, J = 3.5, 0.8 Hz, 6α -H); mass spectrum (12 eV, low resolution), m/e (relative intensity) 490 (M⁺ $C_{29}H_{46}O_6$, 3), 473 (5), 472 (17), 455 (2), 431 (15), 430 (42), 413 (34), 412 (100), 402 (10), 401 (18), 400 (20), 399 (16), 397 (10), 395 (14), 394 (35), 385 (12), 384 (21), 365 (24), 326 (21), 304 (63), 292 (29), 289 (24), 245 (43), 237 (46), 191 (22), 152 (21); high-resolution mass spectrum, observed m/e (composition, interpretation, calculated millimass) 472.31624 (C₂₉H₄₄O₅, M⁺ – H₂O, 472.31887), 454.31132 (C₂₉H₄₂O₄, M⁺ – 2H₂O, 454.30831), 430.30507 (C₂₇H₄₂O₄, M⁺ – AcOH, 430.308 31), 412.297 52 ($C_{27}H_{40}O_3$, M⁺ – H₂O – AcOH, 412.297 74), 400.293 95 ($C_{26}H_{40}O_3$, M⁺ – AcOH – CH₂O, 400.297 74); 394.285 93 ($C_{27}H_{38}O_2$, M⁺ – AcOH – 2H₂O), 382.287 95 ($C_{26}H_{38}O_2$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 382.287 18), 299.164 88 ($C_{19}H_{32}O_3$, M⁺ – AcOH – H₂O – CH₂O, 300 + CH₂O – AcOH – H₂O – AcOH – AcOH – H₂O – AcOH $-H_2O - AcOH - C_8H_{17}$, 299.164 72), 281.154 33 ($C_{19}H_{21}O_2$, M⁺ - $2H_2O - AcOH - C_8H_{17}$, 281.154 15), 245.11806 ($C_{15}H_{17}O_3$, M⁺ - $AcOH - CH_2O - C_{11}H_{23}$, 245.11777), 227.10701 ($C1_5H_{15}O_2$, M⁺ $-H_2O - AcOH - CH_2O - C_{11}H_{23}$, 227.107 20), 155.085 75 ($C_{11}H_{23}$); 300-MHz ¹H NMR (pyridine-d₅) δ 0.82 (3, s, H-18), 0.87 (9, d, J = 7 Hz, H-21, -26, -27), 1.50 (m, H-20, H-25), 1.94 (1, d, J = 15.0 Hz, 12α -H), 2.01 (3, s, OAc), 2.29 (1, dd, J = 15.0, 4.7 Hz, 12β -H), 2.62 (1, m, H-14), 2.77 (1, dd, J = 14.2, 5.2 Hz, 4α -H), 3.92 (1, d, J = 4.7 Hz, 11 β -H), 4.21 (1, dd, J = 12.6, 5 Hz, H-19 coupled also with OH), 4.50 (1, dd, J = 12.6, 5 Hz, H-19 coupled also with OH), 4.75 (1, m, 3α -H), 5.67 (1, br s, 6α -H), 5.82 (1, s, 5-OH), 5.86 (1, br s, H-7), 6.21 (1, d, 3-OH), 6.55 (1, t, 4, C-19-OH).

Acetylation of 1a. Sterol 1a (1 mg) was reacted with 1 mL of acetic anhydride-pyridine (1:19) at room temperature overnight.

Purification of the product by HPLC using silica gel with petroleum ether-chloroform (1:2) afforded 1b which showed the following spectral properties: IR (neat) 3600 (w), 3500 (s), 1742, 1470, 1370, 1240 cm⁻¹; 100-MHz ¹H NMR (CDCl₃) & 0.51 (3, s, H-18), 0.88 (6, d, J = 7 Hz, H-26, -27), 0.90 (3, d, H-21), 2.01, 2.08, 2.17 (3 each, s, OAc), 3.33 (1, d, 5.0, 11 β -H), 4.10 (1, d, J = 11.5 Hz, H-19), 4.55 (1, d, J = 11.5 Hz, H-19), 5.16 (1, m, 3α -H), 5.31 (2, m, H-6, -7); mass spectrum (70 eV, low resolution), m/e (relative intensity) 514 (3% M⁺ - CH₃COOH), 486 (0.9), 472 (0.7), 471 (1.2), 455 (1.3), 412 (2), 395 (13), 394 (32), 281 (5), 253 (8), 248 (16), 232 (21), 149 (19), 129 (11), 106 (46), 83 (100).

Hydrolysis of 1a. 1a (1.5 mg) was heated with 5% KOH in 95% ethanol (5 mL) under reflux for 30 min, and the reaction mixture was cooled to room temperature, acidified, diluted with water, and then extracted with methylene chloride. The organic phase was dried over MgSO₄, the solvent evaporated in vacuo, and the residue purified by HPLC using a silica gel column and CHCl₃-MeOH (98:2) as eluant to yield pure 1c: IR (neat) 3435 (sh), 3340, 1575, 1482, 1375 cm⁻¹; 270-MHz ¹H NMR (pyridine- d_5) δ 0.86 (6, d, J = 7 Hz, H-26, -27), 0.88 (3, s, H-18), 0.97 (3, d, J= 7 Hz, H-21), 3.04 (1, dd, J = 13.1, 2.7 Hz, 4α -H), 3.90 (1, d, J= 8.7 Hz, H-19), 4.16 (1, d, J = 8.7, H-19), 4.48 (2, m, H-11 and COH), 4.52 (1, m, 3a-H), 5.72 (1, s, H-7); low-resolution mass spectrum (70 eV), m/e (relative intensity) 448 (C₂₇H₄₄O₅, 2%), 431 (5), 430 (13), 415 (1), 413 (3), 412 (8), 401 (4), 383 (6), 317 (11), 305 (12), 299 (10), 430 (13), 227 (13), 213 (12), 211 (12), 209 (12), 205 (17), 193 (22), 182 (20), 167 (24), 163 (29), 161 (24), 159 (26), 149 (49), 147 (43), 137 (55), 121 (76), 109 (90), 95 (86), 81 (100)

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Registry No. 1a, 84473-32-5; 1b, 84473-33-6; 1c, 84473-34-7.

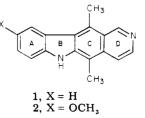
Total Synthesis of Ellipticine and 9-Methoxyellipticine via Benzotriazole Intermediates

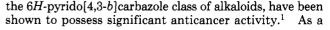
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Ellipticine (1) and 9-methoxyellipticine (2), members of





⁽⁸⁾ Gueran, R. I.; Greenberg, N. H.; MacDonald, M. M.; Schumacher, A. M.; Abbott, B. J. Cancer Chemother. Rep., Part 3, 1972, 3, 1-103. Effective doses (ED₅₀) in the tissue culture tests are expressed as concentrations in $\mu g/mL$ of test material in the growth medium that cause 50% inhibition of cell growth. "Active" materials display in ED₅₀ ≤ 20 $\mu g/mL$. PS refers to in vitro lymphocytic leukemia.